

**ISLAMIC UNIVERSITY OF TECHNOLOGY (IUT)**  
**ORGANISATION OF ISLAMIC COOPERATION (OIC)**  
**DEPARTMENT OF MECHANICAL AND PRODUCTION ENGINEERING**

Semester Final Examination

Winter Semester: 2022 - 2023

Course Number: MCE 4391

Full Marks: 150

Course Title: Basic Mechanical Engineering (EEE)

Time: 3 Hours

There are six questions. Answer all the questions. The symbols have their usual meanings. Marks of each question are mentioned with the questions and corresponding CO and PO and the total marks are written on the right side. Assume reasonable value of missing data.

1. a) A refrigerator uses refrigerant-134a as the working fluid and operates on an ideal vapor-compression refrigeration cycle between 0.2 MPa and 0.5 MPa. The mass flow rate of the refrigerant is 0.08 kg/s. 25  
(CO2)  
(PO2)
- (i) Illustrate the system diagram along with P-h diagram. (3 marks)
  - (ii) Briefly explain the function of the components of the system. (5 marks)
  - Determine the followings:
    - (iii) Enthalpy at all points. (4 marks)
    - (iv) The rate of heat removal from the refrigerated space. (2 marks)
    - (v) The power input to the compressor. (2 marks)
    - (vi) The rate of heat rejection to the environment. (2 marks)
    - (vii) The COP of the refrigerator. (2 marks)
- b) Derive mathematical relation between COP of refrigeration and heat pump system (5 marks)
2. a) (i) Describe how steam is formed and illustrate the effect of increasing and decreasing pressure on the process using P-v diagram. Identify different lines and points from the P-v diagram. (5 marks) 25  
(CO4)  
(PO1)
- (ii) Find the properties of steam mentioned below, under the following conditions using the steam table:
- a. Saturation temperature at 160 kPa. (2 mark)
  - b. Saturation pressure at 150°C. (1 mark)
  - c. At saturated condition, T=120°C and h = 2500 kJ/Kg find the dryness fraction. (2 marks)
  - d. At superheated condition, P=16 MPa and T=600°C find the enthalpy. (2 mark)

- b) (i) Discuss the working principle of boiler through the illustration of simple diagram. (5 marks)  
 (ii) Write a short note on boiler mountings and accessories. (4 marks)  
 (iii) Explain any two of the following terms briefly: (4 marks)
  - Superheater.
  - Water level indicator.
  - Manhole.
  - Economizer

3. a) (i) Define vapor pressure and cavitation. (4 marks) 25  
(CO5)  
 (ii) Classify different types of fluid flow. (5 marks) (PO1)  
 (iii) Differentiate between Newtonian fluid and non-Newtonian fluid. (3 marks) (PO2)  
 (iv) Explain the reason why viscosity of liquids decreases, and the viscosity of gases increases with temperature. (3 marks)
- b) A piezometer and a Pitot tube are tapped into a 40-mm diameter horizontal water pipe, and the height of the water columns are measured to be 0.26 m in the piezometer and 350 mm in the Pitot tube (both measured from the top surface of the pipe).
  - Draw the setup described above. (2 marks)
  - Determine the velocity at the center of the pipe. (3 marks)
- c) The pressure difference between an oil pipe and waterpipe is measured by a double-fluid manometer, as shown in Fig.1. For the given fluid heights and specific gravities, calculate the pressure difference  $\Delta P = P_B - P_A$ . (5 marks)

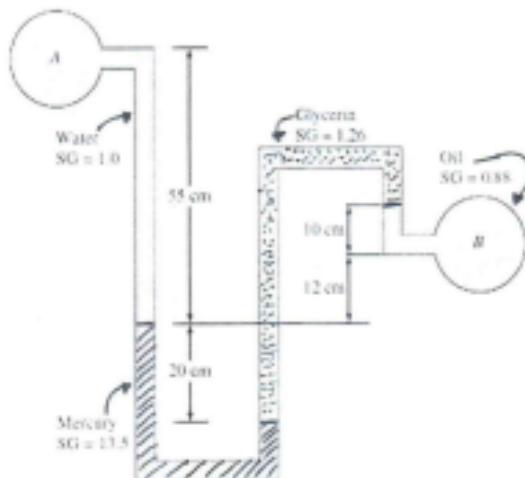


Fig.1

4	a)	<b>Illustrate the construction details of an IC engine.</b> (4 marks)	25
	b)	<b>Illustrate the engine terminologies in a single diagram and explain them briefly.</b> (8 marks)	(CO3) (PO1)
	c)	(i) <b>Illustrate and explain the working principle of a two-stroke engine.</b> (7 marks) (ii) <b>Illustrate and explain briefly, any three from the following terms:</b> (6 marks) <ul style="list-style-type: none"> <li>a. Cylinder block and Cylinder liner</li> <li>b. Piston and Crankcase</li> <li>c. Connecting Rod and Crankshaft</li> <li>d. Intake and exhaust valve, Cylinder Head</li> </ul>	
5	a)	(i) <b>Define priming and its necessity in centrifugal pump.</b> (3 marks) (ii) <b>Differentiate between pump, fan, blower, and compressor in a tabular format.</b> (4 marks)	25 (CO5) (PO1)
	b)	<b>Illustrate the components of a hydroelectric powerplant and explain briefly.</b> (6 marks)	
	c)	(i) <b>State a side-by-side comparison between impulse and reaction turbine.</b> (4 marks) (ii) <b>Illustrate a Pelton wheel, Francis and Kaplan turbine with proper labelling and explain them briefly.</b> (8 marks)	
6	a)	In a piston cylinder arrangement, a gas of with volume $0.192 \text{ m}^3$ is compressed with compression ratio of 13.5 using the law $p v^{1.37} = C$ . At the beginning of the compression, the state of working fluid is 1 bar and 316 K. $[R=289 \text{ J/kg.K}, C_p = 0.996 \text{ kJ/kg.K}]$ <ul style="list-style-type: none"> <li>a. <b>Identify and illustrate the process through P-v diagram.</b> (1 marks)</li> <li>b. <b>Determine the mass of the working fluid.</b> (1 marks)</li> <li>c. <b>Determine temperature, pressure, and volume at the end of the compression.</b> (3 marks)</li> </ul>	25 (CO2) (PO2)
	b)	<b>State and elaborate the Clausius and Kelvin-Planck statements and discuss the perpetual motion machine of second kind based on those statements.</b> (5 marks)	
	c)	An ideal otto cycle has a compression ratio of 8 using 1 kg of air. At the beginning of the compression process, air is at 100 kPa and $17^\circ\text{C}$ . 800 kJ/kg heat is transferred to air during the constant-volume heat addition process. (Take $R=0.287 \text{ kJ/kg.K}, C_p = 1.001 \text{ kJ/kg.K}$ and $C_v = 0.707 \text{ kJ/kg.K}$ ) <ul style="list-style-type: none"> <li>(i) <b>Illustrate the P-v and T-s diagram of the above-mentioned cycle and identify the processes.</b> (4 marks)</li> <li>(ii) <b>Find T, P and V at each state.</b> (8 marks)</li> <li>(iii) <b>Calculate the <math>W_{in}</math>.</b> (2 marks)</li> <li>(iv) <b>Calculate the thermal efficiency of the system.</b> (1 marks)</li> </ul>	

# Saturated Water-Pressure Table

**TABLE A-3**

Saturated water—Pressure table

Press., P kPa	Specific volume, m <sup>3</sup> /kg			Internal energy, kJ/kg			Enthalpy, kJ/kg			Entropy, kJ/kg·K		
	Sat. temp., T <sub>s</sub> , °C	Sat. liquid, v <sub>f</sub>	Sat. vapor, v <sub>g</sub>	Sat. liquid, u <sub>f</sub>	Sat. Evap., u <sub>g</sub>	Sat. vapor, u <sub>g</sub>	Sat. liquid, h <sub>f</sub>	Sat. Evap., h <sub>g</sub>	Sat. vapor, h <sub>g</sub>	Sat. liquid, s <sub>f</sub>	Sat. Evap., s <sub>g</sub>	Sat. vapor, s <sub>g</sub>
20	60.06	0.001017	7.6481	251.40	2204.6	2456.0	251.42	2357.5	2608.9	0.8320	7.0752	7.9073
25	64.96	0.001020	6.2034	271.93	2190.4	2462.4	271.96	2345.5	2617.5	0.8932	6.9310	7.8302
30	69.09	0.001022	5.2287	289.24	2178.5	2467.7	289.27	2335.3	2624.6	0.9441	6.8234	7.7675
40	75.86	0.001026	3.9933	317.58	2158.8	2476.3	317.62	2318.4	2636.1	1.0261	6.6430	7.6691
50	81.32	0.001030	3.2403	340.49	2142.7	2483.2	340.54	2304.7	2645.2	1.0912	6.5019	7.5931
75	91.76	0.001037	2.2172	384.36	2111.8	2896.1	384.44	2278.0	2662.4	1.2132	6.2426	7.4558
100	99.61	0.001043	1.6941	417.40	2088.2	2506.6	417.51	2257.5	2675.0	1.3028	6.0562	7.3589
101.325	99.97	0.001043	1.6734	418.95	2087.0	2506.0	419.06	2256.5	2675.6	1.3069	6.0476	7.3545
125	105.97	0.001048	1.3750	444.23	2068.8	2513.0	444.36	2240.6	2584.9	1.3741	5.9100	7.2841
150	111.35	0.001053	1.1594	466.97	2052.3	2519.2	467.13	2226.0	2693.1	1.4337	5.7894	7.2231
175	116.04	0.001057	1.0037	486.82	2037.7	2524.5	487.01	2213.1	2700.2	1.4850	5.6865	7.1716
200	120.21	0.001061	0.88578	504.50	2024.6	2529.1	504.71	2201.6	2706.3	1.5302	5.5968	7.1270
225	123.97	0.001064	0.79329	520.47	2012.7	2533.2	520.71	2191.0	2711.7	1.5706	5.5171	7.0877
250	127.41	0.001067	0.71873	535.08	2001.8	2536.8	535.35	2181.2	2716.5	1.6072	5.4453	7.0525
275	130.58	0.001070	0.65732	548.57	1991.6	2540.1	548.86	2172.0	2720.9	1.6408	5.3800	7.0207

# Saturated Water-Temperature Table

**TABLE A-4**

Saturated water—Temperature table

Temp., T °C	Specific volume, m <sup>3</sup> /kg			Internal energy, kJ/kg			Enthalpy, kJ/kg			Entropy, kJ/kg·K		
	Sat. press., P <sub>s</sub> , kPa	Sat. liquid, v <sub>f</sub>	Sat. vapor, v <sub>g</sub>	Sat. liquid, u <sub>f</sub>	Sat. Evap., u <sub>g</sub>	Sat. vapor, u <sub>g</sub>	Sat. liquid, h <sub>f</sub>	Sat. Evap., h <sub>g</sub>	Sat. vapor, h <sub>g</sub>	Sat. liquid, s <sub>f</sub>	Sat. Evap., s <sub>g</sub>	Sat. vapor, s <sub>g</sub>
100	101.42	0.001043	1.6720	419.06	2087.0	2506.0	419.17	2256.4	2675.6	1.3072	6.0470	7.3542
105	120.90	0.001047	1.4186	440.15	2071.8	2511.9	440.28	2243.1	2683.4	1.3634	5.9119	7.2952
110	143.38	0.001052	1.2094	461.27	2056.4	2517.7	461.42	2229.7	2691.1	1.4188	5.8193	7.2382
115	169.18	0.001056	1.0360	482.42	2040.9	2523.3	482.59	2216.0	2708.6	1.4737	5.7092	7.1829
120	198.67	0.001060	0.89133	508.60	2025.3	2528.9	503.81	2202.1	2706.0	1.5279	5.6013	7.1292
125	232.23	0.001065	0.77012	524.83	2009.5	2534.3	525.07	2188.1	2713.1	1.5816	5.4956	7.0771
130	270.28	0.001070	0.66808	546.10	1983.4	2539.5	546.38	2173.7	2720.1	1.6346	5.3919	7.0265
135	313.22	0.001075	0.58179	567.41	1977.3	2544.7	567.75	2159.1	2726.9	1.6872	5.2901	6.9773
140	361.53	0.001080	0.50850	588.77	1960.9	2549.6	588.16	2144.3	2733.5	1.7392	5.1901	6.9294
145	415.68	0.001085	0.44600	610.19	1944.2	2554.4	610.64	2129.2	2739.8	1.7908	5.0919	6.8827
150	476.16	0.001091	0.39248	631.66	1927.4	2559.1	632.18	2113.8	2745.9	1.8418	4.9953	6.8371
155	543.49	0.001096	0.34648	653.19	1910.3	2563.5	653.79	2098.0	2751.8	1.8924	4.9002	6.7927
160	618.23	0.001102	0.30580	674.79	1893.0	2567.8	675.47	2082.0	2757.5	1.9426	4.8066	6.7492
165	700.93	0.001108	0.27244	696.46	1875.4	2571.9	697.24	2065.6	2762.8	1.9923	4.7143	6.7067
170	792.18	0.001114	0.24260	718.20	1857.5	2575.7	719.08	2048.8	2767.9	2.0417	4.6233	6.6650

# Superheated Water

**TABLE A-6**

Superheated water (Concluded)

T °C	v m³/kg	u kJ/kg	h kJ/kg	s kJ/kg K	v m³/kg	u kJ/kg	h kJ/kg	s kJ/kg K	v m³/kg	u kJ/kg	h kJ/kg	s kJ/kg K
<i>P = 15.0 MPa (342.16°C)</i>												
Sat.	0.010341	2455.7	2610.8	5.3108	0.007932	2390.7	2529.5	5.1435	0.005862	2294.8	2412.1	4.9310
350	0.011481	2520.9	2693.1	5.4438	0.012463	2684.3	2902.4	5.7211	0.009956	2617.9	2816.9	5.5526
400	0.015671	2740.6	2975.7	5.8819	0.015204	2845.4	3111.4	6.0212	0.012721	2807.3	3061.7	5.9643
450	0.018477	2880.8	3157.9	6.1434	0.017385	2972.4	3276.7	6.2424	0.014793	2945.3	3241.2	6.1446
500	0.020828	2998.4	3310.8	6.3480	0.019305	3085.8	3423.6	6.4266	0.016571	3064.7	3396.2	6.3390
550	0.022945	3106.7	3450.4	6.5730	0.021073	3192.5	3561.3	6.5890	0.018185	3175.3	3539.0	6.5075
600	0.024921	3209.3	3583.1	6.6796	0.022742	3295.8	3693.8	6.7366	0.019695	3281.4	3675.3	6.6593
650	0.026804	3310.1	3712.1	6.8213	0.024342	3317.5	3823.5	6.8715	0.021134	3385.1	3807.8	6.7991
700	0.028621	3409.8	3839.1	6.9573	0.027405	3599.7	4079.3	7.1237	0.023870	3590.1	4067.5	7.0531
800	0.032121	3609.3	4091.1	7.2037	0.030348	3803.5	4334.6	7.3511	0.026484	3795.7	4325.4	7.2829
900	0.035503	3811.2	4343.7	7.4288	0.033215	4010.7	4592.0	7.5616	0.029070	4004.3	4584.7	7.4960
1000	0.038808	4017.1	4599.2	7.6378	0.036029	4272.3	4852.8	7.7588	0.031508	4216.9	4847.0	7.6933
1100	0.042062	4227.7	4858.6	7.8339	0.038806	4438.5	5117.6	7.9449	0.033952	4433.8	5112.9	7.8802
1200	0.045279	4443.1	5122.3	8.0192	0.041556	4659.2	5386.5	8.1215	0.036371	4655.2	5382.7	8.0574
<i>P = 25.0 MPa</i>												
<i>P = 35.0 MPa</i>												
375	0.001978	1799.9	1849.4	4.0345	0.001792	1738.1	1791.9	3.9313	0.001701	1702.8	1762.4	3.8724
400	0.006005	2428.5	2578.7	5.1400	0.002798	2068.9	2152.8	4.4758	0.002105	1914.9	1988.6	4.2144
425	0.007886	2607.8	2805.0	5.4708	0.005299	2452.9	2611.8	5.1473	0.003434	2253.3	2373.5	4.7751
450	0.009176	2721.2	2950.6	5.6759	0.006737	2618.9	2821.0	5.4422	0.004957	2497.5	2671.0	5.1946
500	0.011143	2887.3	3165.9	5.9643	0.008691	2824.0	3084.8	5.7956	0.006933	2755.3	2997.9	5.6331
550	0.012736	3020.8	3339.2	6.1816	0.010175	2974.5	3279.7	6.0403	0.008348	2925.8	3218.0	5.9093
600	0.014140	3140.0	3493.5	6.3637	0.011485	3103.4	3446.8	6.2373	0.009523	3055.6	3399.0	6.1229
650	0.015430	3251.9	3637.7	6.5243	0.012590	3221.7	3599.4	6.4074	0.010565	3190.9	3560.7	6.3030
700	0.016643	3359.9	3776.0	6.6702	0.013654	3314.3	3742.9	6.5599	0.011523	3308.3	3711.6	6.4673
800	0.018922	3570.7	4043.8	6.9322	0.015628	3551.2	4020.0	6.8301	0.013278	3531.6	3996.3	6.7409
900	0.021075	3780.2	4307.1	7.1668	0.017473	3764.6	4288.8	7.0695	0.014904	3749.0	4270.6	6.9853